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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/566,066	06/29/2006	Fritz Gestermann	PO-8729/LcA 36,752	2036
23416 7590 01/12/2009 CONNOLLY BOVE LODGE & HUTZ, LLP P O BOX 2207 WILMINGTON, DE 19899			EXAMINER BELL, BRUCE F	
			ART UNIT 1795	PAPER NUMBER
			MAIL DATE 01/12/2009	DELIVERY MODE PAPER

**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

<b>Office Action Summary</b>	<b>Application No.</b> 10/566,066	<b>Applicant(s)</b> GESTERMANN ET AL.	
	<b>Examiner</b> Bruce F. Bell	<b>Art Unit</b> 1795	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

### Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

### Status

- 1) ☐ Responsive to communication(s) filed on \_\_\_\_.
- 2a) ☒ This action is **FINAL**.                      2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

### Disposition of Claims

- 4) ☒ Claim(s) 12, 13, 15, 16, 18, 19, 21, 22, 24, 25, 27, 29, 31, 33 and 35 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 12, 13, 15, 16, 18, 19, 21, 22, 24, 25, 27, 29, 31, 33 and 35 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_ are subject to restriction and/or election requirement.

### Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

### Priority under 35 U.S.C. § 119

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All    b) ☐ Some \*    c) ☐ None of:
1. ☒ Certified copies of the priority documents have been received.
  2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_.
  3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

### Attachment(s)

- |  |   |
|--|---|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892)                                | 4) <input type="checkbox"/> Interview Summary (PTO-413)<br>Paper No(s)/Mail Date. ____. |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948)                       | 5) <input type="checkbox"/> Notice of Informal Patent Application                       |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08)<br>Paper No(s)/Mail Date ____. | 6) <input type="checkbox"/> Other: ____.  |

## DETAILED ACTION

### ***Claim Rejections - 35 USC § 103***

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

1. Claims 12-13 and 31 are rejected under 35 U.S.C. 103(a) as being unpatentable over DeNora (EP 0785294 A).

DeNora et al disclose an electrochemical cell for the electrolysis of an aqueous solution of hydrogen chloride (col. 4, lines 34-36), at least consisting of one anode half-cell having an anode (col. 4, lines 37-40), a cathode half-cell having a gas diffusion electrode as cathode (col. 4, lines 55-57) and an ion exchange membrane (col. 4, lines 47-50) which is arranged between anode half-cell and cathode half-cell, which ion exchange membrane consists at least of one perfluorosulphonic acid polymer (col. 4, lines 50-53), the gas diffusion electrode and the ion exchange membrane lying adjacently (col. 5, lines 25-28). The DeNora patent further sets forth pressure and temperature conditions in the electrochemical cell by which the ion exchange membrane and the gas diffusion electrode are intimately linked (i.e. having a contact area greater than 50% of the geometric area). The conditions are as follows:

Pressure: 0.1 bar (i.e. 102-1020 g/cm<sup>2</sup>), col. 5, lines 35-37.

Temperature: not above 60<sup>0</sup> C (col. 8, line 9 and example).

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The subject matter as a whole would have been obvious to one of ordinary skill in the art at the time the instant invention was made because even though the prior art of DeNora et al does not disclose the specific contact area being greater than 50%, DeNora et al does disclose intimate contact of those surfaces. Since applicant has not shown data comparing the prior art device with that of the instant invention it appears that intimate does in fact yield a contact region of greater than 50%. Therefore, the prior art of DeNora et al renders the applicants instant invention obvious for the reasons set forth above.

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 12-13, and 31 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kolouch et al (6042702).

Kolouch et al discloses an electrochemical cell which has a perfluorinated cationic membrane that is made of hydrated copolymers of PTFE and PSFF vinyl either containing a pendant of sulfonic acid groups (NAFION) that has an equivalent wt. of 1100 gms., and 1500 gms., wherein the Nafion is a two layer structure. See col. 5, line 61 - col. 6, line 11. The anode and cathode are made of porous, gas diffusion electrodes having a high specific surface area. The anode and cathode comprise an electrochemically active material disposed adjacent, to the surface of the cation

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transporting membrane. A thin film of electrochemically active material may be applied directly to the membrane or in the membrane. The electrochemically active material may comprise a catalytic or metallic material or metallic oxide as long as the material can support charge transfer. The electrochemically active material may be a catalyst material of a noble metal or a transition metal and oxides, alloys or mixtures thereof. The catalyst material may be on a support of carbon and particles of PTFE. The electrode are hot pressed into the membrane in order to have a good contact between the catalyst material and the membrane. See col. 6, line 61 - col. 7, line 34.

The subject matter as a whole would have been obvious to one of ordinary skill in the art at the time the instant invention was made because even though the prior art of Kolouch et al does not specifically address the percentage of contact area between the cathode and ion exchange membrane it appears that the hot pressing method would yield a contact area of the magnitude set forth in applicants instant claims, absent evidence to the contrary.

### ***Claim Rejections - 35 USC § 103***

4. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

5. Claims 12, 13, 15, 16, 31, 33, 35 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shimamune et al (5766429).

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Shimamune et al disclose an electrolytic cell that is partitioned by an ion exchange membrane into anode and cathode compartments wherein at least one of an anode and cathode are closely contacted to the ion exchange membrane to form a gas diffusion electrode. See abstract. A gas diffusion electrode is applied to an ion exchange membrane which gas diffusion layer of the gas diffusion electrode is made hydrophobic (water proofed). A hydrophilic layer is applied to the gas diffusion layer to make the gas diffusion electrode which layer is the reaction layer. The patent further sets forth that by attaining a closely contacting or adhering the ion exchange membrane to the gas diffusion electrode, that a reduction of the bath voltage is achieved and that the electric resistance is minimized. See col. 1, lines 22-63. An ion exchange materials of Nafion is shown in example 1.

The prior art of Shimamune et al does not disclose the contact area of at least 50% of the geometric area.

The subject matter as a whole would have been obvious to one of ordinary skill in the art at the time the instant invention was made because even though Shimamune et al does not set forth the specific contact area of at least 50% of the geometric area, Shimamune et al does disclose a reduction in bath voltage and does disclose that the electric resistance is minimized which is what the applicants instant invention is doing. Therefore, one having ordinary skill in the art would recognize based on the teachings in Shimamune about the closely contacting or adhering ion exchange membrane to the gas diffusion electrode, in conjunction with low electric resistance (indicative of good physical contact) and low bath voltages (also indicative of low electric resistance), that

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in fact, the gas diffusion electrode and the ion exchange membrane must be in contact with each other at an amount of greater than 70% since low electric resistance would not be found at lower percentages. Therefore, the prior art of Shimamune et al renders the applicants instant invention obvious for the reasons set forth above.

The process limitations of pressure and temperature are not given any patentable weight in a product claim unless it can be shown by comparative results between the instant invention and the prior art that such pressures and temperatures materially affect the overall product having the same overall structure.

***Claim Rejections - 35 USC § 103***

6. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

7. Claims 18, 19, 21, 22, 24, 25, 27, 29 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shimamune et al (5766429) in combination with Ford (4242184).

Shimamune et al (5766429) is as disclosed above in the 35 USC 103 rejection.

Shimamune et al does not disclose a dual layer ion exchange membrane.

Ford discloses a membrane for an electrolytic cell having a dual layer ion exchange membrane having different equivalent weights of between 900 and 1600. The dual layers are laminated together to form one membrane. See col. 4, lines 41-46; col. 6, lines 6-22.

The subject matter as a whole would have been obvious to one of ordinary skill in the art at the time the instant invention was made because even though Shimamune does not teach the dual membrane structure or equivalent weights, the prior art of Ford shows the use of such membranes in a similar electrolytic cell, where the membrane of this construction is used to maximize the current efficiency of the electrolytic production through the use of the lower equivalent weight side of the polymer being placed near the side of the anode and the high equivalent weight on the side of the cathode. Therefore, it would be within the ability of the person having ordinary skill in the art to use such a membrane in the Shimamune et al device to optimize the operation of the cell.

***Claim Rejections - 35 USC § 103***

8. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

9. Claims 12, 13, 15, 16, 18, 19, 21, 22, 24, 25, 27, 29, 31, 33, 35 are rejected under 35 U.S.C. 103(a) as being unpatentable over Yoshida et al (4526663).

Yoshida et al disclose an electrolysis cell having anode compartments and cathode compartments separated by a cation exchange membrane, wherein the cathodes are maintained in intimate contact with one entire surface of the cation exchange membrane. See abstract. Yoshida et al further discloses the use of a dual layer ion exchange membrane where one layer has an equivalent weight of 1500 and the other



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side has an equivalent weight of 1100. See example 8. The two ion exchange membrane layers are thermally fused into a two layer laminate. See example 5.

Yoshida et al does not specifically state that the contact area between the gas diffusion electrode and the ion exchange membrane is greater than 50%.

The subject matter as a whole would have been obvious to one of ordinary skill in the art at the time the instant invention was made because even though the prior art of Yoshida et al does not disclose the degree of contact as set forth in the instant claim, the patent does set forth that the cathode and ion exchange membrane are fused and in intimate contact with one another and that the process is done by lamination processing which is known to use pressure. Therefore, absent evidence to the contrary, it appears that a cathode/membrane made as set forth in the Yoshida et al patent, would in fact yield a contact area, between the cathode and ion exchange membrane, greater than 70%. Yoshida et al further discloses the use of a support embedded between the two layers. See example 5. Therefore, the prior art of Yoshida et al renders the applicants instant invention obvious for the reasons set forth above.

### ***Response to Arguments***

10. Applicant's arguments filed 9/30/08 have been fully considered but they are not persuasive.

Applicants argue the rejection of claims 11-13 and 30-31 under 35 USC 102 (b) with respect to both DeNora and Kolouch et al saying that these patents do not teach at least 50% contact area between the geometric areas of the surfaces. The examiner

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disagrees in that intimate contact means that the surfaces are in contact with one another from one end to the other otherwise, it would not be an intimate contact. Further, this is a product claim and as such the process limitations with respect to pressure and temperature have been given little or no patentable weight since it appears that the final product is the same as the instant invention, absent evidence to the contrary. Applicants arguments with respect to the amount of contact in the DeNora and Kolouch patents is here say since applicant has not provided data showing that the instant product is any different than that of the patented product. Applicant argues that their instant data shows that the operating voltages are significantly lower, however applicants have failed to compare these results with those of the prior arts of DeNora and Kolouch showing that these two inventions don't meet those qualifications also. Applicants also argue that the two claimed surfaces are smooth such that the contact are is achieved under operating conditions of the cell. The examiner would like to point out that process limitations are not given any patentable weight in an apparatus or product claim unless it can be shown by comparative results that the final apparatus or product is materially different by using such process, therefore the pressures and temperatures have not been given patentable weight. Therefore, the rejection is maintained.

The rejection of claims 14-29 and 32-35 under 35 USC 103 as being unpatentable over Kolouch in view of Murphy (6059943) has been dropped in view of applicants comments, however, a new rejection has been made above.

***Conclusion***

11. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Bruce F. Bell whose telephone number is 571-272-1296. The examiner can normally be reached on Monday-Friday 6:30 AM - 3:00 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached on 571 272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

BFB  
January 8, 2009

/Bruce F. Bell/  
Primary Examiner, Art Unit 1795